# NATIONAL UNIVERSITY BANGLADESH



First Year Syllabus

Department of Physics

Four Year B.Sc. (Honours) Program

**Effective from the Session: 2024-25** 

#### Mission

To promote "Progress for Everyone" by fostering transformative education, conducting pioneering research, and encouraging meaningful collaboration with individuals, communities, and partners both in Bangladesh and around the world.

#### Vision

We expect that the hundredth anniversary of the National University, our commitment to "Progress for Everyone" will be evident in all areas of our operations and stakeholder engagements.

We leverage our core strength in education, research, and community engagement to address pressing global issues and foster sustainable development.

By 2042, National University aims to strengthen its positions as Bangladesh's top academic institution and rank among the world's leading universities. Our focus will be on expanding expertise building strategic partnerships, and increasing our global influence. Our services will be measured by the achievements of the individuals and communities we serve.

#### **Mission of the Program:**

The academic program is focused on the creation, translation, and dissemination of knowledge on the subject matter. The strategic goals of the department are to:

- 1. Support the aims and objectives of the University within the capacity of our Departmental program. Curriculum for BSc (Hons) in Physics,
- 2. Train and produce high-quality graduates to meet up national and international requirements in scientific sectors of the job market.
- 3. Enhance the teaching-learning and research capacities of the Department by retaining and recruiting outstanding faculty and staff.

#### **Program Educational Objectives (PEO):**

**PEO-1:** To prepare physics graduates to exhibit quality of excellence, critical thinking, creativity, inventiveness, and self-motivation for life-long learning to handle all kind of diverse situations in interdisciplinary and multidisciplinary environment.

**PEO-2:** To produce graduates who are globally acceptable professionals for government, corporate and research organizations along with skills for entrepreneurial pursuits in multidisciplinary areas.

**PEO-3:** To groom graduates who can demonstrate technical competence in the field of project and technology management.

**PEO-4:** To produce graduates who can ethically lead and work as a part of team towards the fulfillment of both individual and organizational goals.

# **Program Learning Outcomes (PLO):**

On successful completion of these programs a student should able to learn the following outcomes:

#### **Fundamental Skill:**

**PLO-1: Knowledge:** Apply fundamental knowledge of physics along with specializations to solve scientific problems of material science, energy, mechanics, optics, communication, space, biomedical physics etc.

PLO-2: Problem analysis: Solve problems in Physics using appropriate mathematical tools.

**PLO-3: Experimentation:** An ability to execute and analyze the results of an experimental investigation or theoretical modeling and to draw valid conclusions with an estimate of the uncertainty in the result. An ability to compare experimental results with the predictions of relevant theories.

**PLO-4:** Conduct investigations of problems: Use research based methods including design of experiments, analysis and interpretation of data leading to logical conclusions.

**PLO-5: Modern tool usage:** Create, select, and apply appropriate techniques, resources, software and modern scientific and ICT tools to problem solving in different sectors of Physics.

#### Social skill:

**PLO-6: Communication:** An ability to communicate scientific information verbally and in the form of clear and accurate scientific reports.

**PLO-7: Environment and sustainability:** Understand the impact of the professional scientific solutions in the societal and environmental contexts, and demonstrate the knowledge of, and the need for sustainable developments.

**PLO-8: Ethics:** Apply ethical principles and commit to professional ethics, responsibilities and norms of scientific practices.

#### Personal Skill:

**PLO-9: Individual and teamwork:** Function effectively as an individual and as a member or leader in diverse teams, and in multidisciplinary settings.

**PLO-10:** Life-long learning: Recognize the need for, and have the preparation and ability to engage in independent and life-long learning in the broadest context of technological change.

#### **Thinking Skill:**

**PLO-11: Project management:** Demonstrate knowledge and understanding of principles of physics and apply those to one's own work as a member and leader in a team, to manage projects and in multidisciplinary environments.

**PLO-12: Technology management:** Competence in using programming languages and in troubleshooting and solving basic problems in electronic equipment and circuits.

# **Mapping PEO with PLO:**

Domain	PLO	PEO-1	PEO-2	PEO-3	PEO-4
	PLO-1	$\sqrt{}$			
C	PLO-2	V	V		
Cognitive	PLO-3			$\sqrt{}$	
nitiv	PLO-4			$\sqrt{}$	
/e	PLO-5			V	
	PLO-6			$\sqrt{}$	
	PLO-7				$\sqrt{}$
Aff	PLO-8				$\sqrt{}$
Affective	PLO-9				
ve	PLO-10				V
Psycho	PLO-11			1	
Psychomotor	PLO-12			V	

# **Courses and Marks Distribution**

# First Year

Course Code	Course Title	Marks	Credits
212701	Mechanics	100	4
212703	Properties of Matter, Waves & Oscillations	100	4
212705	Heat, Thermodynamics and Radiation	100	4
212706	Practical-I (Properties of Matter, Waves & Oscillations, Heat, Thermodynamics and Radiation)	100	4
213709	Foundations of Mathematics	100	4
213711	Calculus-1	50	2
212807	Chemistry-I	100	4
212808	Chemistry-I Practical	50	2
	Or		
213607	Fundamentals of Statistics	100	4
213610	Lab-1: Fundamentals of Statistics	50	2
219901	History of Bangladesh: Language, Culture and Identity	100	4
219903	Information and Communication Technology	75	3
219904	Lab: Information and Communication Technology	25	1
	Total=	900	36

# Matrix of Courses with Program Learning Outcomes (Curriculum map):

	PLO-1	PLO-2	PLO-3	PLO-4	PLO-5	PLO-6	PLO-7	PLO-8	PLO-9	PLO10	PLO11	PLO12
212701	√	√										
212703	<b>√</b>	<b>√</b>										
212705	√	<b>√</b>										
212706	$\checkmark$	$\sqrt{}$	$\sqrt{}$	$\sqrt{}$	$\sqrt{}$				V			
213709	$\sqrt{}$	$\checkmark$										
213711	$\sqrt{}$	$\checkmark$										
212807	$\sqrt{}$	$\sqrt{}$					4					
212808	$\sqrt{}$	$\sqrt{}$		$\sqrt{}$	$\sqrt{}$				V			
212607	$\sqrt{}$	$\sqrt{}$									$\sqrt{}$	
212610	$\sqrt{}$	$\sqrt{}$		$\sqrt{}$	1							
219901	$\sqrt{}$	$\checkmark$					)	$\sqrt{}$				
219903	$\sqrt{}$				$\sqrt{}$	7	$\sqrt{}$	$\sqrt{}$	√	$\sqrt{}$		
219904	$\sqrt{}$				1					$\checkmark$		$\checkmark$

# **Detailed Syllabus**

<b>Course Code</b>	212701	Marks: 100	Credits: 4	Class Hours: 60
<b>Course Title:</b>	Mechanics			

## **Course Objectives:**

CLO7

This course reviews the concepts of mechanics learnt at school from a more advanced perspective and goes on to build new concepts. It begins with Newton's Laws of Motion and ends with the Forces and Special Theory of Relativity. Students will also appreciate to learn the Collisions in CM Frame, Gravitation, Rotational Motion and Oscillations. The students will be able to apply the concepts learnt to several real world problems.

#### **Course Learning Outcomes (CLO):**

After going through the course, the student should be able to learn the following outcomes

CLO1		Understand concept of quantities and units
CLO <sub>2</sub>		Grasp the knowledge of one- two- and three- dimensional motion and their
		applications in real life problems
CLO3	(i)	Understand laws of motion and their applications. He / she will learn the concept
		of momentum, angular momentum to apply them to basic problems
	(ii)	Describe how fictitious forces arise in a non-inertial frame
CLO4		Understand work, energy, power efficiency, conversion of energy and Law of
		conservation of energy
CLO5	(i)	Understand the phenomena of collisions and idea about center of mass and
•		laboratory frames and their correlation. Write the expression for center of mass
		of a rigid body and about a given axis of symmetry for different uniform mass
		distributions and
	(ii)	Explain the concept of law of conservation of momentum and Elastic and non
		elastic collision
CLO6		Understand rotational kinematics

Understand the analogy between translational and rotational dynamics, and application of both motions simultaneously in analyzing rolling with slipping

# **Mapping of CLOs with PLOs**

	PLO-1	PLO-2	PLO-3	PLO-4	PLO-5	PLO-6	PLO-7	PLO-8	PLO-9	PLO10	PLO11	PLO12
CLO1	$\sqrt{}$											
CLO2	$\sqrt{}$											
CLO3	$\sqrt{}$											
CLO4	V	$\sqrt{}$										
CLO5	$\sqrt{}$											
CLO6	V	$\sqrt{}$										\
CLO7	1	<b>V</b>							,			

	Topic	Teaching Learning Strategy	CLOs
1	Vector Algebra: Vector and scalar quantities; Vectors and their components, Vector addition and subtraction, Scalar and vector triple products, scalar and vector fields, Vector differentiation and integration, Gradient, Divergence and Curl and their physical significance, Gauss's divergence theorem, Green's theorem and Stoke's theorem, Polar, Spherical and Cylindrical co-ordinates.	Lecture, Assignment	CLO1
2	Concept of Measurement: Different Measurement units, International system of units, Origin of Length mass and time, Conversion of units from one system to another.	Lecture, Assignment	CLO1
3	Particles Motion in one dimension: Concept of motion and frame of reference, Position and displacement, Average velocity and average speed, Instantaneous velocity and speed, Acceleration, Constant acceleration, Equations for motion with constant acceleration, Free-fall acceleration, Equation for free-fall acceleration, Particles of physics and basic structure of atoms and nuclear.	Lecture, Group Discussion, Assignment	CLO2
4	Particles Motion in Two and Three Dimensions: Position and displacement using vectors, Velocity and average velocity, Acceleration and average	Lecture, Group Discussion, Workshop and	CLO2

	acceleration, Equation of motion using vector, Projectile motion, Uniform circular motion.	Assignment	
5	Force and Motion: Newton's laws of motion and their applications, Concept of mass, Force and weight, Frictional forces and Properties of friction, Drag force and terminal speed, Forces of nature.	Lecture, Group Discussion, and Assignment	CLO3
6	Work, Energy and Power: Kinetic and Potential energy, Work done by constant and variable forces, Work-energy theorem, Hooke's law, Work done by a spring force, Work done by weight, Power, Gravitational potential energy, Conservation of energy.	Lecture, Group Discussion, and Assignment	CLO4
7	<b>System of Particles:</b> Center of mass of systems of particles, Center of mass of rigid bodies, Linear momentum of a particle, Linear momentum of a system of particles, Conservation of linear momentum for a system of particles.	Lecture, Assignment	CLO5
8	Collisions of Bodies: Collisions and its classification, Impulse and linear momentum, Elastic and inelastic collision in one dimension, Motions of the center of mass of colliding bodies.	Lecture, Group Discussion, and Assignment	CLO5
9	Rotational Kinematics: Translational and Rotational motion, Angular Position, Angular displacement, Angular Velocity and angular acceleration, Rotation with constant angular acceleration, Relation between linear and angular kinematics of a particles in circular motion.	Lecture, and Assignment	CLO6
10	Rotational Dynamics: Torque and angular momentum and their relation, Kinetic energy of rotation and rotational inertia (moment of inertia), Combined Translational and rotational motion of a rigid body, Parallel and perpendicular axes theorems of moment of inertia, calculation of moment of inertia for solids of different shapes, conservation of angular momentum. Relation between angular momentum and torque.	Lecture, and Assignment	CLO7

# List of Books:

Halliday, D., Resnick, R., & Walker, J. (2001). *Fundamentals of physics* (6th ed.). John Willey and Sons, Inc.

Khan A. H., & Ishaq. M. (2017). B.Sc. Hons. *Physics 1st part* (Bengali Version). Ideal Books.

Sears, F.W., Zemansky, M.W., & Young, H.D. (2007). *University physics* (12th ed.). Pearson Addison Wesley.

Spiegel, M.R. (1959). Vector analysis. McGraw-Hill.

<b>Course Code</b>	212703	Marks: 100	Credits: 4	Class Hours: 60
<b>Course Title:</b>	Properties of	f Matter, Waves & O	scillations	

#### **Course Objectives**

This course reviews the concepts of mechanics learnt at school from a more advanced perspective and goes on to build new concepts. It includes Newton's Laws of gravitation, Kepler's laws of motion, Properties of matter and waves. Students will also appreciate the Collisions in CM Frame, Gravitation, Rotational Motion and Oscillations. The students will be able to apply the concepts learnt to several real world problems.

#### **Course Learning Outcomes**

After going through the course, the student should be able to

CLO1: Understand Kepler's laws of planetary motion and their applications.

**CLO2:** Demonstrate deformation of solids and to apply them to basic problems.

CLO3: Apply molecular theory to explain surface tension, cohesive and adhesive forces

CLO4: Understand fluid dynamics, Bernoulli's equation and its application.

**CLO5:** Understand the basic properties of waves and sound waves.

**CLO6:** Describe and analyze SHM and DHM

CLO7: Understand and apply vibrational knowledge in longitudinal and transverse mode

of vibration in rods and air columns

# Mapping of CLOs with PLOs

	PLO-1	PLO-2	PLO-3	PLO-4	PLO-5	PLO-6	PLO-7	PLO-8	PLO-9	PLO10	PLO11	PLO12
CLO1	1	$\sqrt{}$										
CLO2	1	1										
CLO3	$\sqrt{}$	$\sqrt{}$										
CLO4	$\sqrt{}$	$\sqrt{}$										
CLO5	$\sqrt{}$	<b>√</b>										
CLO6	$\sqrt{}$	<b>√</b>										
CLO7	$\sqrt{}$	$\sqrt{}$					·					

	Торіс		ng Learning trategy	CLOs
1	Gravitational potential and field in simple Gravitational potential energy.	universal tial and y and its ravity by endulum, e cases,	Lecture, Group Discussion, and Assignment	CLO1
2	Elasticity: Hooke's Law, Elastic constants of solids, Poisson's ratio and their interrelations, elastic potential energy, Experimental determinelastic constants, Torsion of a cylinder, Berbeams, Cantilever, Variation of elasticit temperature.	Internal nation of nding of	Lecture, Group Discussion, Workshop and Assignment	CLO2
3	Surface Tension: Surface tension and surface Adhesive and cohesive forces, Molecular the surface tension, Pressure on a curved member uniform tension, Soap bubble, Capillarity, Accontact and its measurement, Determination of tension of water and mercury drop, Variation of tension with temperature.	neory of brane of Angle of f surface	Lecture, Group Discussion, and Assignment	CLO3
4	Fluid Dynamics: General concepts of fluid Streamlines, Equation of continuity, Be equation, Application of Bernoulli's equate equation of continuity. Coefficient of viscosity, velocity and Reynold's number, Poiseuille's and its correction, Measurement of viscosity, Vof viscosity with temperature.	rnoulli's ion and , Critical formula	Lecture, Group Discussion, and Assignment	CLO4
5	Waves: Waves and Particles, Types of Transverse and Longitudinal waves, Waveler frequency, The Speed of a traveling Wave, War on a stretched string, Energy and power of a string wave, The principle of superposition for	ngth and ve speed traveling	Lecture, Group Discussion, and Assignment	CLO5

	Interference of waves, Complex waves, Standing waves		
	and Resonance.		
6	Sound Waves: The Speed of Sound, Propagation and	Lecture,	CLO5
	speed of longitudinal waves, Traveling longitudinal	Group	
	waves, Standing longitudinal waves, Beats, Doppler	Discussion, and	
	effect.	Assignment	
7	Oscillations: Simple harmonic motion (SHM), Energy	Lecture,	CLO6
	consideration in SHM, Applications of SHM, Relation	Group	
	between SHM and uniform circular motion,	Discussion, and	
	Combinations of two SHM's, Lissajous' figures, Two-	Assignment	
	body oscillations, Damped harmonic motion, Forced		
	oscillations and resonance, Power and intensity of wave		
	motion.		
8	Vibrations: Vibrations of string, Membranes, bars,	Lecture,	CLO7
	plates and air-column, Sonometer, Melde's experiment,	Group	
	Rectangular and circular membranes, Transverse and	Discussion, and	
	longitudinal vibration of rod, Air-columns in cylindrical	Assignment	
	pipes, Organ pipes, Chladni's figure.		

# List of Books:

Halliday, D., Resnick, R., & Walker, J. (2001). Fundamentals of physics (6th ed.). John Willey and Sons, Inc.

Khan, A.H., & Ishaq, M. (2017). Physics 2<sup>nd</sup> Part (Bengali Version). Ideal Books.

Sears, F.W., Zemansky, M.W., & Young, H.D. (2007). *University physics* (12 th ed.). Pearson Addison Wesley.

<b>Course Code</b>	212705	Marks: 100	Credits: 4	Class Hours: 60
<b>Course Title:</b>	Heat, The	rmodynamics &	Radiation	

#### **Course Objectives:**

This course deals with the relationship between the macroscopic properties of physical systems in equilibrium. It reviews the concepts of thermodynamics learnt at school from a more advanced perspective and develops them further. The primary goal is to understand the fundamental laws of thermodynamics and their applications to various systems and processes. In addition, it will also give exposure to students about the Kinetic theory of gases, transport phenomena involved in ideal gases, phase transitions and behavior of real gases.

#### **Course Learning Outcomes:**

After completing the course student will be able to

- CLO1: Understand basic concept of heat, temperature, and thermal equilibrium
  - (ii) Construct different types of thermometer
- **CLO2:** Understand kinetic theory of gas
- CLO3: (i)Comprehend the basic concepts of thermodynamics, the first and the second law of thermodynamics, the concept of entropy and the associated theorems, the thermodynamic potentials and their physical interpretations.
  - (ii) Learn about Maxwell's thermodynamic relations.
  - (iii) Understand the basic aspects of kinetic theory of gases, Maxwell-Boltzman distribution law, equitation of energies, mean free path of molecular collisions, viscosity, thermal conductivity, diffusion and Brownian motion.
- **CLO4:** Explain the mechanism of heat engine and Carnot's heat cycle.
- CLO5: Understand real gas equations, Van der Waal equation of state, the Joule-Thompson effect.
- CLO6: Describe the concept of black body radiation, Emissive and absorptive powers,
  Stefan- Boltzmann's Law, Wien's displacement law, Rayleigh-Jean's law,
  Planck's quantum hypothesis and applications of radiations laws.

# **Mapping of CLOs with PLOs**

	PLO-1	PLO-2	PLO-3	PLO-4	PLO-5	PLO-6	PLO-7	PLO-8	PLO-9	PLO10	PLO11	PLO12
CLO1	$\sqrt{}$	$\sqrt{}$										
CLO2	1	$\sqrt{}$										
CLO3	<b>V</b>	<b>V</b>										
CLO4	1	$\sqrt{}$										
CLO5	<b>V</b>	V										\
CLO6	<b>V</b>	V										

	Topics	Teaching Learning Strategy	CLOs
1	Heat and Temperature: Concept of temperature, Thermal equilibrium, Measurement of low and high temperatures, The Clausius & Fahrenheit scales, Thermal expansion, Gas Thermometers, Platinum resistance thermometer, Thermocouple	Lecture, Group Discussion, and Assignment	CLO1
2	Kinetic Theory of Gases: Equation of state of an ideal gas, Equipartition of energy, Translational kinetic energy, Mean free path, Maxwell's theory of distribution of velocities, Brownian motion, Degrees of freedom & Molar Specific Heats, Van der Waals' Equation of state, Transport phenomena.	Lecture, Group Discussion, and Assignment	CLO2
3	<b>Transmission of Heat:</b> Conduction, Convection, Radiation, Conduction of heat in solids, Measurement of thermal conductivity of a bad conductor, Heat conduction through composite walls.	Lecture, Group Discussion, and Assignment	CLO3
4	First Law of Thermodynamics: Internal Energy, Heat and work, Isothermal and adiabatic processes, Work done by expanding gases, Statement of first law of thermodynamics and applications.	Lecture, Group Discussion, and Assignment	CLO3

5	Second Law of Thermodynamics and Entropy: The	Lecture,	CLO4
	Thermodynamic temperature scale, Concept of entropy,	Group	
	Calculation of 6entropy change in reversible and	Discussion,	
	irreversible processes, Entropy and second law of	Assignment	
	thermodynamics, Entropy and disorder. The Carnot		
	engine, Efficiency of heat engines, Carnot's theorem,		
	Refrigerator and air-conditioner, Clausius theorem,		
	Clausius-Clapeyron equation.		
6	Third Law of Thermodynamics: Nernst heat theorem,	Lecture,	CLO3
	Phase rule and its uses, Third law of thermodynamics.	Group	
	, , , , , , , , , , , , , , , , , , ,	Discussion, and	
_		Assignment	CT OF
7	Thermodynamic Functions: Thermodynamic	Lecture,	CLO5
	potentials at constant volume and pressure, Maxwell's	Group Discussion, and	
	thermodynamic relations, Specific heat equations,	Assignment	
		rissignment	
	Joule-Thomson effect and its applications.		
8	Radiation Laws: Concept of black body and black body	Lecture,	CLO6
	radiation, Emissive and absorptive powers, Kirchhoff's	Group	
	law, Stefan-Boltzmann's Law, Wien's displacement	Discussion, and Assignment	
		Assignment	
	law, Rayleigh-Jean's law, Planck's quantum hypothesis,		
	Planck's law, Applications of radiations laws.		

# **List of Books**

Halliday, D., Resnick, R., & Walker, J. (2001). Fundamentals of physics (6th ed.). John Willey and Sons, Inc

Hossain, T. (1975) Text Book of heat. Variety Books.

Khan, A.H., & Ishaq, M. (2017). B.Sc. Hons physics 3<sup>rd</sup> Part. (Bengali Version) Ideal Books.

Saha, M.N., & Srivastava, B.N. A (1958) Treatise on heat. The Indian Press Private Ltd.

Sears, F.W. (1950). An Introduction to thermodynamics. Addison-Wesley Press

Zemansky, M.W. (1997) Heat and thermodynamics (7th ed.). The McGraw-Hill Co. Inc.

<b>Course Code</b>	212706	Marks: 100	Credits: 4	Class Hours: 60								
<b>Course Title:</b>	Practical-I	Practical-I (Properties of Matter, Waves & Oscillations, Heat,										
	Thermodyn	amics and Radiation	1)									

#### **Course Objectives:**

Students will able to be familiar with different physical concepts relating to the experiments with complete understanding, realize the applications of these concepts such as mechanics, properties of matter and become confident with the related theories. The aim of this Laboratory is to understand some of the basic phenomenon of mechanics through various experiment. Another prime objective of the course is to enhance the scientific data collection and analysis in Physics Laboratories.

#### Course Learning Outcomes: Students will be able to

- **CLO1:** Understand different quantities and determination of g
- CLO2: (i) Gain the concept Understand measurement techniques of modulus of rigidity.
  - (ii) Gain the concept of modulus of elasticity.
- **CLO3:** (i)Understand and measure the surface tension of liquids which do wet the contact materials and understand the effect of capillarity.
  - (ii) Understand and measure the effect of surface tension of a liquid which does not wet the contact material and learn the concept about the effect of cohesive and adhesive forces.
  - (iii) Learn the concept of surface tension of water, how to form ripples on water surface, and know how to measure the wave length of ripples.
- **CLO4:** Learn the concept of viscosity and capillary flow of water.
- CLO5: Understand the techniques to determine specific heat capacity of solid and liquid
- **CLO6:** Measure thermal conductivity of solid
- **CLO7:** Grasp the clear knowledge of temperature coefficient of resistance
- CLO8: Understand the laws of transverse vibration of a stretched string by experimentation
- **CLO9:** Determine frequency of a tuning fork by Melde's experiment.
- **CLO10:** Determination of latent heat of fusion of ice and latent heat of condensation of steam.

#### Mapping of CLOs with PLOs

	PLO-1	PLO-2	PLO-3	PLO-4	PLO-5	PLO-6	PLO-7	PLO-8	PLO-9	PLO10	PLO11	PLO12
CLO1			<b>V</b>	V								
CLO2			V	V								
CLO3			<b>√</b>	V								
CLO4			1	V								
CLO5			V	V								
CLO6			<b>√</b>	V								
CLO7			<b>√</b>	<b>√</b>								
CLO8			<b>√</b>	V								
CLO9			<b>√</b>	V								
CLO10			1	1								

To perform two experiments (one from each group) each of three hours duration.

i)	Experiments	(3 hours each)	$2 \times 35 =$
1,	LAPCITUCIUS	(5 Hours cach	2 A 33

Total marks = 35 Marks for each experiment shall be distributed as follows:

a) Theory	8
b) Data collection and tabulation	n 12
c) Calculation, graphs and resul	t 10
d) Discussion	5
	Total marks - 25

# **Course Contents (List of Experiments):**

# Group - A

- 1. Determination of acceleration due to gravity 'g' by compound pendulum.
- 2. Determination of acceleration due to gravity 'g' by Katter's pendulum.
- 3. Determination of Young's modulus and rigidity modulus by Searle's dynamic method.
- 4. Determination of rigidity modulus of a wire/rod by static method.
- 5. Determination of rigidity modulus of the material of a wire by dynamic method.

- 6. Determination of the spring constant and effective mass of a given spiral spring and hence to calculate the rigidity modulus of the material of the spring.
- 7. Determination of the Young's modulus by the flexure of a beam (bending method).
- 8. Determination of the moment of inertia of a fly-wheel about its axis of rotation.
- 9. Determination of the Young's modulus for the material of a wire by Searle's apparatus.
- 10. Determination of Surface tension of water by capillary tube method.
- 11. Determination of surface tension of mercury by Quicken's method.

#### Group - B

- 12. Determination of the specific heat of solid by method of mixture, with radiation correction.
- 13. Determination of the specific heat of a liquid by the method of cooling.
- 14. Determination of the thermal conductivity of a good conductor by Searle's apparatus.
- 15. Determination of the thermal conductivity of a bad conductor by Lee's method.
- 16. Determination of mechanical equivalent of heat 'J' with radiation correction.
- 17. Investigation of the variation of resistance of a copper wire with temperature and determination of its temperature coefficient of resistance.
- 18. Verify the laws of transverse vibration of a stretched string with a sonometer (n-l), and n-1/l curves only)
- 19. Determination of the frequency of a tuning fork by Melde's experiment.
- 20. Determination of latent heat of fusion of ice with radiation correction.
- 21. Determination of latent heat of condensation of steam with radiation correction.
- 22. Determination of density of water at various temperature by specific gravity bottle and study the variation of density with temperature from the graph.

#### **List of Books:**

Ahamed, G.U., & Uddin, M.S. (1969). Practical physics (4th ed.). Hafiz Book Centre.

Din, K., & Matin, M.A. (1966). Advanced practical physic. Mallik Brothers.

Khan, A.H. (2019). B.Sc. Hons. practical physics. Ideal Books.

Worsnop B.I,. & Flint, H.T. (1957). Advanced practical physics. Methuen and Co. Ltd.

Course Code	213709	<b>Marks: 100</b>	Credits: 4	Class Hours: 60

<b>Course Title:</b>	Foundations of Mathematics

# **Course Objectives:**

The emphasis of course is to equip students with the mathematical and critical skills required in solving problems of interest to physicists. The course will also expose students to fundamental computational physics skills enabling them to solve a wide range of physics problems. The skills developed during course will prepare them not only for doing fundamental and applied but also for a wide variety of careers.

# **Course Learning Outcomes**

CLO1	Understand the number system and theory of equations which have applications in
	problems
CLO2	Describe relations between roots and coefficients, Symmetric functions of roots
CLO3	Learn the System of linear equations and basic properties of matrices, different types
	of matrices viz., Hermitian, skew Hermitian, orthogonal and unitary matrices and
	their correspondence to physical quantities.
CLO4	Vector Spaces, Linear transformations, Matrix representation of linear
	transformation, Kernel and image, Eigenvalues and Eigenvectors
CLO5	Explain vector spaces and 2D and 3D geometry.

## Mapping of CLOs with PLOs

	PLO-1	PLO-2	PLO-3	PLO-4	PLO-5	PLO-6	PLO-7	PLO-8	PLO-9	PLO10	PLO11	PLO12
CLO1		V									$\sqrt{}$	$\sqrt{}$
CLO2		V									1	1
CLO3		$\checkmark$									$\sqrt{}$	$\sqrt{}$
CLO4		$\sqrt{}$									$\sqrt{}$	$\sqrt{}$
CLO5		1				•			·		1	1

Topic	Teaching	CLOs
	<b>Learning Strategy</b>	
Relations and functions: Relations; Order relation;	Whiteboard-based	
Equivalence relations; Functions; Images and inverse	lecture	
images of sets; Injective, surjective, and bijective	Assignment	CLO1
functions; Inverse functions.	Short presentations	
	by students	
Real Number System: Field and order properties; Prime	Whiteboard-based	
numbers; Natural numbers; Integers and rational numbers;	lecture	
Absolute value and its properties; Basic inequalities	Assignment	CLO2
(including inequalities of means and powers); Inequalities		
of Cauchy, Chebyshev, Weierstrass.		
Complex Number System: Field of Complex numbers;	Whiteboard-based	
De Moivre's theorem and its applications.	lecture	CLO2
	Assignment	
Matrices and Determinants: Notion of matrix; Algebra	Whiteboard-based	
of matrices; Different types of matrices; Invertible	lecture	
matrices; Determinant function; Properties of	Assignment	
determinants; Minors, Cofactors, expansion, and		CLO3
evaluation of determinants. Elementary row and column		
operations and row-reduced echelon matrices, Invertible		
matrices, Diagonal, triangular, and Symmetric matrices.		
System of Linear Equations: Linear equations; System of	Whiteboard-based	
linear equations (homogeneous and non-homogeneous)	lecture	
and their solutions using different methods. Gaussian	Assignment	CLO3
elimination, Application of matrices and determinants for		CLOS
solving systems of linear equations, Applications of systems		
of equations in real-life problems.		
Summation of finite series: Arithmetic and geometric	Whiteboard-based	
series; Method of difference; Successive differences;	lecture	CLO4
Summation of trigonometric series.	Assignment	CLOT
Theory of Equations: Relations between roots and	Whiteboard-based	
coefficients; Symmetric functions of roots; Sum of the	lecture	CLO3
powers of roots; Synthetic division; Descartes' rule of	Assignment	
signs; Multiplicity of roots; Transformation of equation.		
Two-dimensional Geometry: Transformation of		CLO5
coordinates, Pair of straight lines (homogeneous second-	lecture	CLOS

degree equations, General second-degree equations representing pair of straight lines, angle between pair of straight lines, Bisectors of angle between pair of straight lines), General equations of second degree (reduction to standard forms, Identifications, Properties and tracing of conics).	Assignment	
Three-dimensional Geometry: Three-dimensional	Whiteboard-based	
coordinates, Distance, Direction cosines and direction		CLO5
ratios, Planes and straight lines.	Assignment	\
Vector Geometry: Vectors in plane and space; Algebra of	Whiteboard-based	CLO6
vectors; Rectangular components; Scalar and Vector	lecture	CLO7
products; Coplanar vectors; Scalar triple product and	Assignment	
vector triple product; Applications of vectors to geometry		
(vector equations of straight lines, planes, areas and		
volumes).		
Vector Spaces: Euclidean <i>n</i> -space, Real vector spaces,	Whiteboard-based	CLO6
Subspaces, Linear combination of vectors, Linear	lecture	CLO7
dependence of vectors, Basis and dimension, Linear	Assignment	
transformations, Matrix representation of linear		
transformation, Kernel and image, Eigenvalues and		
Eigenvectors.		

#### **List of Books:**

Anton, H., & Rorres, C. (2013). Elementary linear algebra with applications (11th ed.). Wiley.

Bernet, Cristofer T. (2017). Geometry (6th ed.). Schaum's Out line, McGraw Hill.

Bernard, S., & Child, J. M. (2016). *Higher algebra* (6th ed.). Arihant Prakashan.

Hall, H. S., & Knight, S. R. (1992). Higher algebra (7th ed.). Macmillan.

Howard A., & Chris R., (2015). Elementary linear algebra with supplemental application. Wiley.

Hummel, J. A. (1994). Vector geometry. Prentice Hall.

Khosh M. (2010). Analytic geometry and vector analysis. Dhaka: Ideal Library.

Lipschutz, S. (1997). Schaum's outline of set theory and related topics (Schaum's Outline Series). McGraw-Hill.

Murrey, R., Spiegel, Seymore L., & Dennis S. (2009). *Vector analysis*, Schaum's Out line, Springer.

Rahman, M. A. (2015). *Basic algebra*. Dhaka: Nahar Book Depot and Publications.

Rahman, M. A. (2016). Linear algebra. Dhaka: Nahar Book Depot and Publications.

Spiegel, M. R. (1974). Vector analysis (Schaum's Outline Series). McGraw-Hill.

<b>Course Code</b>	213711	Marks: 50	Credits: 2	Class Hours: 30
<b>Course Title:</b>	Calculus- I			

## **Course Description**

Calculus is one of the most fundamental courses in Mathematics, which consists of two parts (Differential and Integral). The course is designed to develop the topics of Differential and Integral calculus. Understanding this course will enable everyone to learn other mathematical courses that require the fundamentals of differentiation and integration.

#### **Objective of the Course**

To develop the basic ideas of functions and their graphs. Learning the basic properties of limit and continuity and analyzing them both mathematically and graphically. To understand the ideas and applications in solving real-life-oriented problems of differentiation and integration.

#### **Course Learning Outcomes**

After completing this course, students will be able to:

- CLO1 Identify and graph various types of functions, including polynomial, rational, exponential, logarithmic, trigonometric, and hyperbolic functions, and describe their key properties.
- **CLO2** Explain the concepts of limits and continuity, and apply relevant theorems to compute limits.
- CLO3 Apply techniques of differentiation, including rules and theorems (e.g., Leibniz's rule), to solve problems related to rates of change and approximations in mathematical and real-life contexts.
- **CLO4** Analyze functions using derivative-based tools, such as the Mean Value Theorem, to determine maximum and minimum values, and concavity, to solve optimization and curve analysis problems.
- **CLO5** Apply techniques of integration (definite and indefinite) and related theorems to solve problems involving area, volume, arc length, and surface area.
- **CLO6** Apply approximation techniques using Taylor polynomials and series to estimate function values and analyze convergence.
- **CLO7** Evaluate series expansions and perform differentiation and integration of series to solve complex calculus problems.

# Mapping of CLOs with PLOs

	PLO-1	PLO-2	PLO-3	PLO-4	PLO-5	PLO-6	PLO-7	PLO-8	PLO-9	PLO10	PLO11	PLO12
CLO1		V									1	
CLO2		V									1	
CLO3		V									V	
CLO4		V									V	
CLO5		V									1	
CLO6		1				·					1	
CLO7		$\sqrt{}$									1	

Topic	Teaching Learning	CLOs
	Strategy	
Functions and Their Graphs: Polynomial and	Whiteboard-based lecture	CLO1
rational functions; logarithmic and exponential	Assignment	
functions; trigonometric functions and their inverses;		
hyperbolic functions and their inverses; combinations		
of such functions.		
Limit and Continuity: Definitions and basic	Whiteboard-based lecture,	CLO2
theorems on limit and continuity; Limit at infinity and	Assignment	
infinite limits; Computation of limits.		
<b>Differentiation:</b> Tangent lines and rates of change;	Whiteboard-based lecture	CLO3
Definition of derivative; One-sided derivatives; Rules	Assignment	
of differentiation; Successive differentiation;	Tutorial	
Leibnitz's theorem; Related rates; Linear		
approximations and differentials.		
Applications of Differentiation: Mean value	Whiteboard-based lecture,	CLO4
theorem; Maximum and minimum values of functions;	Assignment	
Concavity and points of inflection; Optimization		
problems.		
<b>Integration:</b> Antiderivatives and indefinite integrals;	Whiteboard-based lecture	CLO5
Techniques of integration; Definite integration using	Assignment	
antiderivatives; Fundamental theorems of calculus;		
Basic properties of integration; Integration by		
reduction.		
<b>Applications of Integration:</b> Arc length; Plane areas;	Whiteboard-based lecture	CLO5
Surfaces of revolution; Volumes of solids of	Assignment	
revolution; Volumes by cylindrical shells; Volumes by		
cross sections.		

Approximation and Series: Taylor polynomials and	Whiteboard-based lecture	CLO6
series; Convergence of series; Taylor's series; Taylor's	Assignment	CLO7
theorem and remainders; Differentiation and		
integration of series.		

#### **List of Books**

- Anton, H., Bivens, I. C., & Davis, S. (2016). *Calculus: Early transcendental* (11th ed.). Wiley. Das, B. C., & Mukherjee, B. N. (1938). *Integral calculus*. Kolkata: U. N. Dhur & Sons Pvt. Ltd.
- Das, B. C., & Mukherjee, B. N. (1949). *Differential calculus*. Kolkata: U. N. Dhur & Sons Pvt. Ltd.
- Matin, M. A., & Chakraborty, B. (1994). *Differential calculus*. Dhaka: Standard Publications. Mohammad, K., Bhattacharjee, P. K., & Latif, M. A. (1968). *Differential calculus* (1st ed.).
- Chittagong: S. Tripaty.
- Mohammad, K., & Bhattacharjee, P. K. (1987). *Integral calculus* (6th ed.). Chittagong: H. Bhattacharjee.
- Stewart, J. (2015). Calculus: Early transcendentals (8th ed.). Cengage Learning.
- Swokowski, E. W. (1988). Calculus with analytic geometry (6th ed.). Brooks/Cole.
- Thomas, G. B., & Finney, R. L. (1996). *Calculus and analytic geometry* (9th ed.). Addison-Wesley.

Course Code 212807	Marks: 100	Credits: 4	Class Hours: 60
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Course Title	Chemistry-I
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## **Course Objective**

The course is designed for the student to impart and improve fundamental knowledge and aspects related to understand the fundamental physical, Organic and Inorganic chemistry topics. To develop and extend student's knowledge on state of matter, structure of atom, radiochemistry, chemical bonding, Liquid and Solution, chemical equilibrium, pH, aliphatic and aromatic hydrocarbons.

#### **Course Learning Outcomes (CLOs)**

After completing this course, students will be able to:

- CLO1 Explain the historical background, fundamental concept, and scope and aim of concepts of physical, Organic and inorganic chemistry.
- **CLO2** Clarify the state of matter, structure of atom, radiochemistry
- **CLO3** Explain about the chemical bonding, Liquid and Solution
- CLO4 explain the chemical equilibrium, pH, aliphatic and aromatic hydrocarbons
- **CLO5** Explain the chemical equilibrium, pH, aliphatic and aromatic hydrocarbons, bonding, synthesis, nomenclature and properties.

#### Mapping of CLOs with PLOs

	PLO-1	PLO-2	PLO-3	PLO-4	PLO-5	PLO-6	PLO-7	PLO-8	PLO-9	PLO10	PLO11	PLO12
CLO1	7	1										
CLO2	7	1										
CLO3	1	1										
CLO4	1	√										
CLO5	<b>V</b>	√										

	Topics	Teaching Learning Strategy	CLOs
1	State of Matter in Chemistry: Nature and classification of matter. Classical states of solid, liquid, gas and Liquid crystals, state of aggregation of matter. Changes of state of matter. Intramolecular and intermolecular force and their role for the determination of state of matter. Modern state as plasma. Avogadro's number with application, Concept of mole: calculation and importance. Stoichiometry.	Lecture, Group Discussion, Assignment	1, 2, 3
2	Structure of atom: Fundamental particles of atoms, Nucleus and discovery of nucleus, atomic model of Rutherford, Bohr atom model: postulates, limitations and success, Spectrum of atomic hydrogen, Dual nature of electron, Heisenberg uncertainty principle, Quantum numbers, atomic orbitals, Aufbau principle, Pauli exclusion principle, Hund's rule of maximum multiplicity, electronic configuration of atoms.	Lecture, Group Discussion, Assignment	3, 4, 5
3	Radiochemistry: Atomic number and mass number, atomic mass unit, atomic nucleus, Isotopes, Nuclear stability, natural and artificial radioactivity, half life and average life of radioelements, radioactive decay, nuclear reactions.	Lecture, Group Discussion, Assignment	3, 4, 5
4	Periodic table: Periodic law, Periodic table, Periodic properties of the elements such as ionization energies, Electron affinity, Electro negativity, Atomic/ionic radius along a period and down a group, Diagonal relationship.	Lecture, Group Discussion, Assignment	2, 3, 4
5	Chemical bond: Definition and causes chemical bonds, octet rule and duplet rule, types of chemical bonds: ionic, covalent bond, polar and non-polar bond, Ionic character of covalent compound and covalent character of ionic compounds VSEPR theory, Valence bond theory, Hybridization, $\sigma$ and $\pi$ -bonding in compounds, Molecular orbital theory, coordination, metallic, hydrogen bonds.	Lecture, Group Discussion, Assignment	2, 3, 4, 5
6	Chemical Reactions: Physical change and Chemical change, Methods of bringing about chemical reaction, Redox reactions, oxidizing and reducing agents, oxidation state and oxidation number, redox half reactions, rules for balancing redox reactions, acid base reactions.	Lecture, Group Discussion, Assignment	2, 3, 4

7	Gaseous State: Vapor and gas, measurable properties of gases, determination of gas pressure, the gas laws: derivation, molecular explanation and practical applications, effusion, diffusion, osmosis, ideal and real gases, ideal gas equation with applications, kinetic theory of gas, deviation from ideal behavior, Van der Walls equation, critical constants. Dalton's law of partial pressure.	Lecture, Group Discussion, Assignment	1, 3
8	Liquids and Solutions: Liquids and their characteristics, evaporation, measurement vapor pressure, dependence of vapor pressure on temperature variation, boiling, distillation, Types of solution, solubility and solubility curve, units of concentration for solution, raoult's law, ideal and non-ideal solutions. Henry's law. a brief treatment of colligative properties.	Lecture, Group Discussion, Assignment	4, 5
9	Chemical Equilibrium: Equilibrium in chemical reactions. Stoichiometry, law of mass action, concentration versus time curves, equilibrium law and equilibrium constant: Kp, Kc, and Kx, effects of temperature, pressure and concentration changes of equilibrium, principle and applications of Le Chatelier and Braun, reaction quotient, extent of reaction, thermodynamic equilibrium constant.	Lecture, Group Discussion, Assignment	1
10	pH and Buffer: Ostwald dilution law, solubility product with principle and applications, common ion effect, ionic product of water (K <sub>W</sub> ), pH with applications, buffer solution, buffer mechanism, Henderson-Hasselbach equation. Acid-base titrations, Acid-base indicators. Theories of acids and bases, conjugate acids and bases, neutralization reactions, acid-base strength, Salts: Classification and their applications, neutralization reactions.	Lecture, Group Discussion, Assignment	1, 2
11	Aliphatic hydrocarbons: Organic compounds, homologous series, the causes of diversity of organic compounds, Hydrocarbons: classifications and IUPAC nomenclature. Formation of carbon cations, carbanion, free radicals and their stabilities. Sources, properties, general methods of preparation of following hydrocarbons: Alkane, Alkene and Alkynes. Classification of organic compounds based on their functional groups.	Lecture, Group Discussion, Assignment	5

12	Aromatic hydrocarbons: Definition and classification,		
	nucleus and side chain, Structure, preparation and uses of	Lecture, Group	5
	benzene. Aromaticity, Electrophilic aromatic substitution	Discussion,	
	reactions with reference to nitration, halogenation,	Assignment	
	sulphonation and alkylation. Orientation of substituents in		
	benzene ring, activating and deactivating group.		

#### **List of Books:**

- Bahl, B. S., & Bahl, A. (2024). *Text-Book of organic chemistry*. S. Chand & Company Ltd Ram Nagar, New Delhi
- English, J., & H.G Cassidy and Baird, R. I. (1949). Principles of organic chemistry: An introductory text in organic chemistry. McGraw Hill.
- Finar, I. L. (1973). Organic chemistry, Vol. I. Pearson Education India.
- Griffin, R. W. (2024). Modern organic chemistry. McGraw Hill.
- Morrison, R.T. & Boyd, R. N. (1992). Organic chemistry. Benjamin-Cummings Pub Co
- Muny, J. Mc. (2010). Organic chemistry. Thomson Broks-Coole.
- Robert, J. D., & Caserio, M. C. (1977). *Basic principles of organic chemistry*. W. A. Benjamin, Inc., Menlo Park, CA.
- Solomons, T. W. G. (2008). Fundamental of organic chemistry. Wiley.

<b>Course Code</b>	Course Title	Marks	Credits	Class Hours
212810	Chemistry-I Practical	50	2	30

#### **Course Objectives**

To gain a fundamental knowledge on laboratory safety policy, some inorganic salts and radical analysis, preparation of solution with standardization, titration and precipitation.

## **Course Learning Outcomes (CLOs)**

- **CLO1** Discuss about the general laboratory safety policy, rules and regulations and chemical management process.
- **CLO2** Explain the applications of apparatus and reagents in various experiments, different laboratory glassware and chemicals.
- CLO3 Prepare stander solution and pH- neutralization curves
- **CLO4** Qualitative analysis of mixtures of inorganic salts consisting of up to five different radicals
- **CLO5** Explain to determination of Fe<sup>2+</sup>, copper (II), nickel as Ni (HDMG) 2 complex 7 using different methods.

## Mapping of CLOs with PLOs

	PLO-1	PLO-2	PLO-3	PLO-4	PLO-5	PLO-6	PLO-7	PLO-8	PLO-9	PLO10	PLO11	PLO12
CLO1			1	V	$\sqrt{}$						<b>√</b>	
CLO2			1	<b>√</b>	<b>V</b>						<b>√</b>	
CLO3			<b>V</b>	<b>√</b>	<b>V</b>						<b>√</b>	
CLO4			<b>V</b>	<b>V</b>	<b>V</b>						<b>√</b>	
CLO5			<b>√</b>	<b>V</b>	$\sqrt{}$						<b>√</b>	

	Topics	Teaching Learning Strategy	CLOs
1	<b>Safety:</b> Introduction of chemical, equipment's and safety in the laboratory.	Lecture, Group Discussion, Experimental, Assignment	1-5
2	<b>Preparation:</b> Preparation of FeSO <sub>4</sub> .7H <sub>2</sub> O Mohr's salt and potash alum.	Lecture, Group Discussion, Experimental, Assignment	1-5
3	<b>Separation and identification</b> : Separation and identification of four radicals from a mixture of anions and cations The cations are Pb <sup>2+</sup> , Cu <sup>2+</sup> , Cd <sup>2+</sup> , Al <sup>3+</sup> , Fe <sup>2+</sup> , Fe <sup>3+</sup> , Co <sup>2+</sup> , Ni <sup>2+</sup> , Zn <sup>2+</sup> Ca <sup>2+</sup> , Ba <sup>2+</sup> , Na <sup>+</sup> , K <sup>+</sup> , and NH <sub>4</sub> <sup>+</sup> , the anions are NO <sub>3</sub> <sup>-</sup> , CO <sub>3</sub> <sup>2-</sup> , S <sup>2-</sup> , SO <sub>4</sub> <sup>2-</sup> , Cl <sup>-</sup> , Br <sup>-</sup> and I <sup>-</sup> .	Lecture, Group Discussion, Experimental, Assignment	1-5
4	<b>Standardization:</b> Standardization of NaOH solution using standard oxalic acid solution.	Lecture, Group Discussion, Experimental, Assignment	1-5
5	<b>Determination:</b> Determination of Fe <sup>2+</sup> using standard permanganate solution.	Lecture, Group Discussion, Experimental, Assignment	1-5
6	<b>Iodometric method:</b> Determination of copper (II) using standard Na <sub>2</sub> SO <sub>3</sub> solution.	Lecture, Group Discussion, Experimental, Assignment	1-5
7	Gravimetric Method: Gravimetric determination of nickel as Ni (HDMG) <sub>2</sub> complex.	Lecture, Group Discussion, Experimental, Assignment	1-5
8	<b>pH-Neutralization Curves:</b> Determination of the pH-neutralization curves of a strong acid by a strong base.	Lecture, Group Discussion, Experimental, Assignment	1-5

#### **List of Books**

Alexeyev, V. (2024). Qualitative analysis. Mir Publishers.

Braun, D. R. (2024). Introduction to chemical analysis. McGraw Hill International,

Christian, G. D. (2023). Analytical chemistry. John Wiley & Sons.

Fifield, F. W., & Kealey, D. (2000). Principles & practice of analytical chemistry. Wiley.

Skoog, D. A., & West, M. D. (2022). Fundamental of analytical chemistry. Publisher, Cengage Learning.

Vogel, A. I. (2017). A Text-Book of macro and semi-micro qualitative inorganic analysis. Longmans, Green and Co. Ltd.

Welcher, F. J., & Hahn, R. B. (2024). Semi-micro qualitative analysis. D. Van Nostrand Co. Inc.

Wulfsberg, G. (1987). *Principles of Descriptive inorganic chemistry*. University Science Books, Mill Valley.

Course Code 213607	Marks: 100	Credits: 4	Class Hours: 60
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<b>Course Title</b>	Fundamentals of Statistics

## **Course Objectives**

To be able to understand the nature, characteristics, scope, application and abuse of statistics. To make familiar with data, nature of data, how to process and condense the data, sources of data and graphical presentation of data, and to apply appropriate statistical tools and techniques to analyze the data. To acquaint students with necessary skills for solving probability related problems using appropriate laws. To provide knowledge on time series and statistical indices.

## **Course Learning Outcomes**

- CLO1 Understand the nature, characteristics, scope, application and abuse of statistics. Knowledge on sources of data and how to process, condense and presentation of data. Understand different characteristics of statistical data such as measures of location, dispersion, moments, skewness, kurtosis and their properties.
- CLO2 Understand the relationship between variables such as simple correlation, rank correlation, correlation ratio, simple regression analysis. standard error of estimators of regression coefficients & their properties and fitting of regression lines.
- CLO3 Comprehend different approaches of defining probability and useful laws of probability to solve problems. Also learn some commonly used probability distributions.
- **CLO4** Gain knowledge about meaning and application of statistical indices.
- CLO5 Identify the pattern and trends and isolate the influencing factors of the time series data for future planning and control.
- CLO6 Understand the concept of numerical methods, including interpolation and use of numerical methods in application to real problems.

## Mapping of CLOs with PLOs

	PLO-1	PLO-2	PLO-3	PLO-4	PLO-5	PLO-6	PLO-7	PLO-8	PLO-9	PLO10	PLO11	PLO12
CLO1			1	$\checkmark$	$\checkmark$						$\checkmark$	
CLO2			$\sqrt{}$	$\sqrt{}$	$\sqrt{}$						$\checkmark$	
CLO3			1	1	$\checkmark$						1	
CLO4			$\checkmark$	$\checkmark$	$\checkmark$						$\checkmark$	
CLO5			$\sqrt{}$	$\checkmark$	$\checkmark$						$\checkmark$	
CLO6			$\checkmark$	$\checkmark$	$\checkmark$						$\checkmark$	
CLO7			$\sqrt{}$	$\sqrt{}$	$\sqrt{}$						$\sqrt{}$	

Торіс	Teaching Learning Strategy	CLOs
<b>Descriptive Statistics:</b> Statistics—Its nature and some important uses,	Lecture,	CLO1
Qualitative and quantitative data, Classification, Tabulation and	Assignment	
frequency distribution, Graphical representation of data, Measures of		
location, Measures of Dispersion, Skewness and Kurtosis,		
Mathematical relationship among different measures of location,		
dispersion, Skewness and kurtosis.		
Bivariate Data: Correlation coefficient, Correlation analysis, The	Lecture,	CLO2
purpose and uses of regression analysis, Simple regression and	Assignment	
methods of least squares and estimation of parameters, Correlation		
ratio, Rank correlation, Partial and multiple correlation.		
Elementary Probability: Meaning of Probability, Classical and	Lecture, Group	CLO3
empirical definitions of Probability, Axiomatic approach of defining	Discussion, Assignment	
probability, Event, Sample space and simple problems on probability,	rissignment	
Addition rule, Conditional probability, Multiplication rule and Bayes		
theorems, The concept of a random variables, Probability function		
and probability density function, Joint probability function. Marginal		
and conditional distributions, Statistical independence, Expected		
value and related theorems, Moment generating function, Common		
probability distributions, Binomial, Poisson and Normal.		
Index Number: Concept of an index number and problems in the	Lecture, Group	CLO4
construction of index number, Types of indices (Price, Quantity,	Discussion, Workshop and	
Value and cost of living indices) and their uses, Tests for index	Assignment	
numbers.		
Time Series analysis: Elements of time-series analysis,	Lecture, Group	CLO5
Measurement of trend by moving average, By least square method,	Discussion, and	
Trend curve, Determination of seasonal indices, Cyclical	Assignment	
movements.		
Numerical Mathematics: Differences of a polynomial, Finite	Lecture, Group Discussion,	CLO6

difference	operator,	Difference	table,	Newton's	formula	and	Workshop and	
starling's co	entral differ	rence formula	a, Invers	se interpolat	ion, Nume	rical	Assignment	
integration				-				
integration	•							

#### List of Books

Gupta, S.C., & Kapoor, V.K. (1994). Fundamentals of applied statistics. Sultan Chand & Sons.

Islam, M.N. (2015). An introduction to statistics and probability (4th ed.). Mullick & Brothers.

Jalil, M. A. and Ferdous R. (1999). *Basic statistics: Methods and applications*, Robi Mostafa, M.G. (1989). *Method of statistics (4<sup>th</sup> ed.)*. Karim press and Publications.

Shil, R.N., & Debnuth, S.C. (2016). *An introduction to the theory of statistics*. Star Publications.

Weiss, N. A., & C. A. Weiss (2012). *Introductory statistics*. Pearson Education.

<b>Course Code</b>	213610	Marks:50	Credits: 2	Class Hours: 30
<b>Course Title:</b>	Lab-1: Fund	lamentals of Statisti	cs	

Data condensation and tabulation. Formation of frequency distribution from both qualitative and quantitative data. Construction of bivariate table. Graphical representation of data. Measures of location and dispersion, Calculation of moments, Measures of skewness and kurtosis. Simple correlation coefficient and fitting of regression lines. Computation of rank correlation coefficient. Fitting of Binomial, Normal and Poisson's distributions, Finding trend values and seasonal variation from time series data by different methods, Calculation of Index numbers and test of index number, Use of Newton's forward and backward formula, Solution of numerical integration.

<b>Course Code</b>	219901	Marks: 100	Credits: 4
Course Title	History of Bangladesh;	Hours 60	

#### **Course Objectives**

This course is designed to help undergraduate students from diverse academic backgrounds develop a comprehensive and nuanced understanding of the historical development of Bangladesh, with a particular focus on its language, culture, and identity. By critically examining a variety of historical events, socio-political movements, and cultural shifts from ancient times to the contemporary period, students will have the opportunity to trace the evolution of the Bengali nation. The course aims to foster informed citizenship through an exploration of the Liberation War, identity formation processes, cultural heritage, the lives and contributions of key political figures, and the roles of ethnic minorities. Ultimately, the objective is to equip learners with the intellectual tools necessary to contextualize current national debates and to recognize the role of youth and globalization in shaping the future of Bangladesh.

### **Course Learning Outcomes (CLOs)**

At the end of the course, learners will be able to:

**CLO 1:** Recall key historical events, dates, movements, personalities, and cultural developments that contributed to the emergence of Bangladesh from the pre-colonial period to the present day.

**CLO 2:** Demonstrate an understanding of the socio-political, linguistic, and cultural influences that shaped Bengali identity and nationalism, emphasizing the Language Movement, Liberation War, and subsequent democratic uprisings.

**CLO 3:** Apply their historical and cultural understanding to interpret contemporary national issues and debates relating to identity, political reform, and cultural transformation in Bangladesh.

**CLO 4**: Analyze how historical events, political ideologies, and social changes are interconnected in shaping the current socio-political landscape of Bangladesh, with a particular emphasis on the roles of movements, minorities, and the youth.

### **Course Contents**

Course	Contents	T	1
	Course contents	Teaching Learning Strategy	CLOs
1. Pre-	colonial Era		
•	Life and Culture of the People in Ancient Bengal		
•	Bengal Under Muslim Rulers: Society, Culture, and Religion	Interactive	CLO 1-
•	Role of the Sufis in Preaching Islam and Impacts of	lectures	2
•	Sufism on the Bengali Society		
•	Bengali Society and Culture in the Writings of Foreigners		
	onial Era (18th and 19th Century)		
•	The Battle of Plassey (1757) and the Beginning of British	T	
	Colonialism	Interactive	CLO 1-
•	Bengal Renaissance	lectures, reading, and	2
•	Reforms in Hindu Society	assignments	2
•	New Forms in Bengali Literature and Culture	assignments	
•	The Muslim Response to Western Education		
3. Colo	onial Era (First Half of the 20th Century)		
•	Partition of Bengal (1905)		
•	Hindu-Muslim Disagreements		
•	Formation of the All India Muslim League (AIML)	T4	
•	Muslim Shahitya Samaj	Interactive lectures and	CLO 2-
•	Buddhir Mukti Andolan: The Urge for Rational Thinking in		3
	Bengali Muslim Society	group discussions	3
•	Growth of Religion-based Identity	discussions	
•	Politics of Hindutva and the Two-Nation Theory		
•	Spread of Communalism in Society		
•	The Partition of India and Bengal		
4. Post	-Partition Era <u>(</u> 1947-1971)		
	Language Movement	Interactive	CI O A
	Political, Economic, and Cultural Aspects	lectures and	CLO 2-
•	Growth of Vernacular Nationalism	group	3
•	Cultural Activism	discussion	
5. Cha	nging Bengali Identity		
	The Evolution of Bengali Identity in the Context of		
	Language, Culture, and Religion		
	From Ancient Times to the Present		
	The Role of the Bengali Language in Shaping Identity	Interactive	GY C -
•	The Language Movement of 1952 and Its Long-term Impact	lectures and	CLO 2-
	on National	thematic	3
	Consciousness	assignments	
	The Influence of Religion on Bengali Identity		
	The Impact of Socio-political Movements on Identity		
	Formation		

	***		
The Liberation			
1 -	Debates on Bengali Identity		
• The Role of Yo	outh in Redefining Identity		
• The Influence	of Globalization		
6. Liberation Movem	ent of 1971 and Mass Uprising till 2024		
	nd Economic Exploitation of East Pakistan by		
West Pakistan			
	Movement, Uprising of 1969		
• The 1970 Gene	. 1		
	eration Movement and the Declaration of		
Independence	ration wo venient and the Declaration of		
-	K Fazlul Haque, Huseyn Shaheed	Interactive	
	Iaulana Bhashani, Sheikh Mujibur Rahman	lectures,	
and Ziaur Rahi		group	CLO 1
The Liberation		discussions,	CLO 3-
Genocide and l		and thematic	4
	e Mukti Bahini	assignments	
	ence Challenges		
Nation-buildin			
Political Instab			
	s and Democratic Movements		
	eracy Movement of the 1980s		
	vement for Democracy and Afterwards		
	ements for Political and Social Reforms		
7. History of Other E			
_	s Communities of Bangladesh	T.,4.,4.	
Historical Pres		Interactive	
Cultural Practic		lectures and	CLO 4
		group discussions	CLO 4
Contributions a	d Cultural Assimilation	discussions	
	sentation of Ethnic Minorities		
_	and Modern Transformations		
	of Bengali Culture		
• From Ancient	Traditions to Modern Expressions	Interactive	
• The Role of Li	terature, Music, and Art in Shaping Bengali	lectures,	
Identity		documentary	
	of Rabindranath Tagore, Kazi Nazrul Islam,	screening,	CLO 1-
and Other Cult	_	and thematic	4
• The Impact of	Globalization on Bengali Culture	assignments	
*	Traditional Arts		
• The Role of Yo	outh in Cultural Innovation Urbanization		
• Environmental			
Liiviioiiiiciitai	Changes		

The Commodification of Culture	

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<b>Course Code</b>	219903	Marks: 75	Credits: 3
<b>Course Title:</b>	Information and Con	nmunication Technolog	y

## **Course Objectives**

The main objective of the course is to develop students' understanding and skills in using, managing, and applying technology to solve problems and enhance various aspects of life and work. Key areas of focus include understanding ICT systems, software, hardware, networks, and their applications.

## **Course Learning Outcomes**

CLO 1	Explain foundational ICT concepts, including the information processing cycle.
CLO 2	Operate standard computer hardware and software systems effectively.
CLO 3	Use office productivity tools (Word, Excel, PowerPoint) for academic and professional tasks.
CLO 4	Apply safe internet practices and use internet tools for communication and information retrieval.
CLO 5	Understand and explain the concepts of Data Analytics, Artificial Intelligence (AI), and Machine Learning (ML)

### **Course Contents**

Unit	Specific Objectives	Content	Teaching and Learning Approach
Unit 1: Introduction to Information and Communications Technology (ICT) and Computer System	<ul> <li>Explain the concept of ICT and its related terminologies</li> <li>Describe the information processing cycle</li> <li>Analyze the impact of ICT on educational, social and economic development</li> <li>Identify career opportunities in ICT</li> </ul>	Information  Application of ICT, Advantages and Disadvantages of ICT, Impact of ICT Career opportunities in ICT education  Information Processing Cycle. Classification of Computers	Classroom Lectures
	education	The Vital Components of	

		1 ~ ~~.	<u> </u>
	<ul> <li>Differentiate among the classes of computers and the usages of them</li> <li>Identify the vital components of the Systems Unit</li> </ul>	the Systems Unit.	
Unit 2: Computer Hardware and Software	<ul> <li>The student will be able to:</li> <li>Describe the categories of Computer Hardware</li> <li>Describe the commonly used Input and Output devices</li> <li>Identify the main processing devices, storage devices and media. Identify the main communication devices.</li> <li>Identify types of Software packages.</li> <li>Distinguish between an Operating System and Application software.</li> <li>Identify different types, examples and uses of Operating Systems and Application software</li> <li>Distinguish between Open Source and Proprietary Software</li> </ul>	Categories of Computer Hardware: Input devices, Processing devices, Output devices, Storage devices  Communication devices, Main Processing Devices: The Processor, Control Unit and Arithmetic and Logic Unit  Software Packages, Operating Systems, Types and uses of Operating Systems, Types and uses Application Software: Educational software, Games software, Graphics software  Browsers: Internet explorer, Google chrome, Mozilla Firefox, Opera, Internet explorer, Mozilla Firefox; Proprietary and Open Source Software	Classroom lectures and Lab
Unit 3: Introduction to Word Processing Application	<ul> <li>The student will be able to:</li> <li>Identify Word Processing packages</li> <li>Create and save a</li> </ul>	Word Processing Packages and Their Uses Creating a Document Using a Word Processor Saving a Document Using	Classroom Lectures, Lab, and Hands-on Practice
	document using the Word Processor • Format a Word document using formatting tools	the 'Save As' command  Editing a Word Document Using Common Editing Tools: Copy/cut, paste,	

Unit 4:	Demonstrate the ability to perform collaborative editing  Insert tables in a Word Processing document  Insert symbols and pictures in Word Processing documents.  Use layout techniques in document creation.  Inserting headers and footers  Print documents using the various print options.	Undo and redo, find, replace, clipboard  Creating a Document with More Sub-Headings and Paragraphs  Text correction, Wrapping options, Text orientation  Formatting and saving a Word document using the formatting tools: font (style, size, color, etc.), bold, underline, italic, superscript, subscript, shadow, strikeout, font color  Paragraph Editing: alignment, bullet & numbering, indent, line spacing, table border  Collaborative Editing: Using the highlighting option to track changes in a document, accepting or rejecting changes  Insert: Adding text comments, Inserting  Tables in a Word Document and inserting  Symbols and Pictures in a Word Document and inserting  Symbols and Pictures in a Word Document. Header, footer, page number, drop cap, word art  Page Design: watermark, page border  Layout: Page setup/print	Classroom
Spreadsheet Application	<ul> <li>Identify Spreadsheet Packages</li> <li>Explain the importance of the Spreadsheet</li> </ul>	e.g. Excel  Importance of Spreadsheet application in Data Management, Related Concepts and	Lectures, Lab and Hands-on Practice

	application in data	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	
	<ul><li>management</li><li>Explain related concepts</li></ul>	cell(s), rows, columns, worksheet, workbook)	
	and terminologies in the Spreadsheet  Identify features in the Spreadsheet application	Features in the Spreadsheet Window Tool Bars: formatting bar, standard bar, formulae bar	
	window  Create and save a Workbook  Construct and insert simple formulae and functions  Format the worksheet using formatting tools.  Printing a worksheet	Types of Data and Their Uses (e.g., number, date, text, currency), Creating and Saving a Workbook, Constructing and Inserting Simple Formulae and Functions  Formatting Worksheet	
		Using Formatting Tools  Draw a Graph/chart Editing and Printing	
		Worksheet	
Unit 5: Presentation Application	<ul> <li>The student will be able to:</li> <li>State the importance of the Presentation application</li> <li>Identify the commonly used features of a Presentation application</li> </ul>	Presentation Applications Packages, Devices used for Presenting, Importance of Presentation Application, Principles for Designing Presentations,	Classroom Lectures, Lab and Hands-on Practice
	while prepare a presentation  • Create and save presentations using a template, Add new slide(s). Edit text, Format	Terminologies in Presentation  Application (eg, Slide Layout, Slide transitions, Slide show)	
	text, Insert objects, images and pictures, Run slide show, Apply transition, animation effects to slides  • Select the print option for printing	Identification of Commonly Used Features of Presentation Application Window: Toolbars, Different presentation view modes  Presentation	
	Prepare a presentation on a selected topic and present it	Prepare a Presentation: Adding elements and formatting slides, slide Show, slide transition, animation Effects	

Unit 6: Privacy and Security	The student will be able to:  • Understand the basics of digital security	Selection of Print Option: Entire presentation, Specific slides, Handouts, Notes pages, outline view of slides, and Number of copies  Introduction to Information Security, cybercrime, DoS and DDoS Attack, Key	Classroom Lectures, Lab and Hands-on
	<ul> <li>Use some security tools.</li> <li>Understanding digital ethics.</li> </ul>	Management, Digital Signature and Certifications, privacy, Data Security, Vulnerability, Threat and Risk, Malware, Social Engineering, Hacking, Plagiarism, Fishing, Software Piracy, Worms and Viruses, Spam, Adware, Malware, Spyware, Antivirus Software  Ethics in the digital world	Practice
Unit 7: Using The Internet to Communicate and Accessing Information	<ul> <li>The student will be able to:</li> <li>Explain basic concepts, requirements, and terminologies of the Internet</li> <li>Apply the rules and regulations in the use of the internet.</li> <li>Using email</li> <li>Use the internet social network to communicate.</li> <li>Use Uniform Resource Locators (URLs) to access Information. Use search engines to access information</li> <li>Upload files to virtual drives and work on it</li> </ul>	Internet, Intranet, Extranet, IP Address, Masking, MAC Address, Internet Services, OSI Reference Model, TCP/IP	Classroom Lectures, Lab and Hands-on Practice

		A	1
		Account, Sending, Accessing Email Messages, Attaching Documents to Email Messages, Using the Internet: Social Networks to Communicate, Uniform Resource Locators (URLs) to Access Information, Using Search Engines, Downloading Information from the Internet. Transferring Information from the Internet to a Different Application	
		Intellectual Property Rights, ICT Policy, Software Piracy etc.  Using Cloud Space:	
		Google Drive, Google Workspace, OneDrive, Dropbox	
Unit8: Emerging Technologies: Data analytics, Artificial intelligence, Machine learning	<ul> <li>Define Data Analytics,         Artificial Intelligence         (AI), and Machine         Learning (ML)</li> <li>Understand how data is         collected, processed, and         used for decision-         making</li> <li>Recognize the role of AI         and ML in everyday life.</li> <li>Explain basic         differences between AI         and ML</li> <li>Discuss benefits and         challenges of these         technologies</li> <li>Recognize current trends</li> </ul>	Introduction to Data Analytics: What is Data? Types of Data, Basic Steps in Data Analytics, Simple Tools; Artificial Intelligence (AI): Making Machines Capable of Performing Tasks that Require Human-Like Thinking  Common Examples: Voice Assistants, Facial Recognition, GPS Route Suggestions, Chatbots.; AI Capabilities: Understanding Language, Recognizing Patterns,	Classroom Lectures, Lab and Hands-on Practice
	and career opportunities in these fields	Making Decisions; Machine Learning (ML): Define Machine Learning (ML), How It Works,	

Classification and Examples; Relationship Between Data Analytics, AI, and ML	
Benefits & Challenges of Data Analytics, AI, and ML. Future Trends & Career Paths	

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<b>Course Code</b>	219904	Marks: 25	Credits: 1	Class Hours: 15
Course Title: Lab: Information and Communication Technology				logy

## **Course Objectives:**

The main objective of the course is to teach the students' understanding and skills in using, and applying technology to solve problems and enhance various aspects of life and work. It includes assembling hardware, installing software, preparing PPT slides, and producing Word and Excel documents.

## **Course Learning Outcomes:** The student will be able to:

CLO 1	Learn assembling hardware
CLO 2	Prepare, edit and print word documents and excel
CLO 3	Prepare power point presentation.
CLO 4	Access information from e-mail
CLO 5	Installation of anti-virus software
CLO 6	Data collection, Prediction using AI, ML, Data Analytics

## **List of Experiments**

CLO	Unit	Experiments	Teaching
Addressed			Learning
			Approach
		Assemble different hardware	
CLO1	1-4	Install different software	Lab and Hands-
		Operate the computer - Drive, folder and file	on Practice
		management	
		Maintenance	
CLO 2	3	Word	Lab and Hands-
		• Prepare a Word document on a specific topic	on Practice
		(e.g, routine, question paper, CV, reports,	
		applications)	
		• Formatting the document (Alignment, table,	
		border, watermark, etc.), e.g., newspaper article,	
		academic report, or documentation used in daily	
		life, book, poster	
		• Print documents with different paper and	
		printers	

CLO 2	4	Excel	
		Prepare a grade sheet	Lab and Hands-
		Prepare a family expenditure	on Practice
		• Prepare a business expenditure report	
		• Prepare payroll management, with a report	
		• Create graphs on the given data	
		• Print Excel files	
CLO 3	5	Power point	Lab and Hands-
		• Prepare an academic presentation on a specific topic.	on Practice
		• Formatting the slides & using different tools.	
		Apply animation and transition	
		• Print PPT files in different modes: Hand note,	
		Slides shorter, outline	
CLO4	6	• Install antivirus software, e.g., Norton	Lab and Hands-
		Antivirus, McAfee, Kaspersky, Avast.	on Practice
CLO5	7	• Use of email	Lab and Hands-
		• Access information from the internet, use a	on Practice
		search engine.	
		Use of virtual drive for collaboration	
CLO6	8	Google Meet, Zoom  Description:  Output  Description:	Lab and Hands-
CLO	ð	Data Collection and Visualization	on Practice
		Simple Prediction Using Trendlines	on ractice

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## **Allied Physics Courses for Other Subjects**

<b>Course Code</b>	212707	Marks: 100	Credits: 4	Class Hours: 60	
<b>Course Title</b>	ourse Title Physics-I (Mechanics, Properties of Matter, Waves & Optics)				

# **Course Objectives:**

This course reviews the concepts of mechanics learnt at colleges from a more advanced perspective and goes on to build new concepts. It begins with Newton's Laws of Motion and ends with the Gravitation and Optics. Students will be taught the concept of Work, Energy, Power, Elasticity, Rotational Motion and Surface Tension. The students will be able to apply the concepts learnt to several real world problems.

# **Course Learning Outcomes (CLO):**

Understand concept of Vector quantities and units
Calculate work, energy and power
Understand rotational motion
Understand analyze gravitation
Understand the laws of fluid dynamics
Grasp the knowledge of physical properties of matter
Understand the concept of waves and optics

	Торіс	Teaching Learning Strategy	CLOs
1	Vector Analysis: Vectors and scalars, Addition and	Lecture,	CLO1
	multiplication of vectors, Triple scalar & vector products, Derivatives of vectors, Gradient, divergence and curl-their	and Assignment	
	physical significance, Theorems of Gauss, Green & Stoke's	5	
2	Work Energy and Power: Work energy theorem,	Lecture,	CLO2
	Conservation of energy and linear momentum,	Group	
	Conservative and non-conservative forces and systems,	Discussion,	
	Conservation of energy and momentum, Centre of mass,	and	

	Collision problems	Assignment	
3	Rotational Motions: Rotational variables, Rotation with	Lecture,	CLO3
	constant angular acceleration, Relation between linear and	Group	
	angular kinematics, Torque on a particle, Angular	Discussion,	
	momentum of a particle, kinetic energy of rotation and	and	
	moment of inertia. Combined translational and rotational	Assignment	
	motion of a rigid body, Conservation of angular momentum.		
4	Gravitation: Centre of gravity of extended bodies,	Lecture,	CLO4
	Gravitational field and potential their calculations,	Group	
	Determination of gravitation constant and gravity,	Discussion,	
	Compound and Katter's pendulums, Motion of planets and	and	
	satellites, Escape velocity	Assignment	
5	Fluid Dynamics: Viscosity and coefficient of viscosity, Poiseulle's equation, Determination of the coefficient of viscosity of liquid by Stoke's method, Bernoulli's theorem and its applications, Torricelli 's theorem, Venturimeter	Lecture, Group Discussion, and Assignment	CLO5
6	Elasticity: Moduli of elasticity, Poisson's ratio, Relations	Lecture,	CLO6
	between elastic constants and their determination, Cantilever	and Assignment	
7	Surface Tension: Surface tension as a molecular	Lecture,	CLO6
	phenomenon, Surface tension and surface energy, Capillary rise or fall of liquids, Pressure on a curved membrane due to surface tension, Determination of surface tension of water, mercury and soap solution, Effect of temperature	and Assignment	
8	Waves: Mechanical waves, types of waves, travelling	Lecture,	CLO7
	waves. The superposition principle. Wave speed, Power and	Group Discussion,	
	intensity in wave motion. Interference of waves, Standing	and	
	Waves and resonance	Assignment	
9	Oscillatory Motions: Simple harmonic motion,	Lecture and	CLO7
	Combination of harmonic motions, Damped harmonic	Assignment	
	motion, Forced oscillations and resonance		

Ī	10	Optics: Fresnel's and Fraunhofer types, Diffraction through	Lecture,	CLO7	
		single slit and double slit, diffraction grating, Dispersive	Group		
			Discussion,		
		and resolving powers of gratings. Plane, Elliptical and	and		l
		circular Polarizations, Optical, Rotatory	Assignment		I
П					

Khan A. H., and Ishaq. M. (2017). B.Sc. Hons. *Physics 1st Part* (Bengali Version). Ideal Books. Halliday, D., Resnick, R., and Walker, J. (2001). *Fundamentals of physics*. (6 th ed.). John Willey and Sons, Inc.

Sears, F.W., Zemansky, M.W., & Young, H.D. (2007). *University physics* (12 th ed.). Pearson Addison Wesley.

Spiegel, M.R. (1959). Vector analysis. McGraw-Hill.

<b>Course Code</b>	212709	Marks: 50	Credits: 2	Class Hours: 30
<b>Course Title</b>	e Title Physics-II (Heat, Thermodynamics and Radiation)			

## **Course Learning Outcomes**

**CLO1:** Understand basic concept of heat, temperature, and thermal equilibrium

**CLO2:** Calculate thermal energy by applying knowledge of calorimetry

**CLO3:** Comprehend the basic concepts of thermodynamics, the first and the second law of thermodynamics and 2<sup>nd</sup> law of thermodynamics.

**CLO4:** the concept of entropy and the associated theorems, the thermodynamic potentials and their physical interpretations

**CLO5:** Understand the basic aspects of kinetic theory of gases, Maxwell-Boltzmann distribution law.

CLO6: Describe the concept of black body radiation, Emissive and absorptive powers, Rayleigh-Jean's law and applications of radiations laws.

#### **Course Contents**

	Торіс	Teaching Learning Strategy	CLOs
1	<b>Thermometry:</b> Temperature, Concepts of thermal	Lecture,	CLO1
	equilibrium, measurement of low and high temperature:	and	
	Gas thermometers, Resistance thermometer,	Assignment	
	Thermocouple, Pyrometry, International temperature scale		
2	Calorimetry: Specific heats of solids, liquids and gases by	Lecture	CLO2
	method of mixture with radiation corrections: Newton's	and	
	Law of cooling, Variations of specific heats, Atomic and	Assignment	
	molecular heats		
3	Thermodynamic Systems: Concept of internal energy:	Lecture	CLO3
	The first law of thermodynamics, Work and specific heats,	and	
	Isothermal and adiabatic processes	Assignment	
4	The second law of Thermodynamics: Reversible and	Lecture,	CLO3
	irreversible processes: Carnot cycle, Efficiency of	Group	
	reversible engines, Absolute thermodynamic temperature	Discussion,	
	scale, Change of phase: Clausius and Clapeyron equation,	and	

	Porous plug experiment	Assignment	
5	<b>Entropy:</b> Entropy of an ideal gas, Temperature-entropy diagram, Increase of entropy.	Lecture	CLO4
6	Thermodynamic Functions: The Maxwell's relations, Specific heat equations.	Lecture	CLO5
7	Radiation: Concept of black body radiation, Kirchhoff's law, Stefan–Boltzmann law, Wien's displacement Law, Rayleigh-Jean's law, Planck's Radiation law, Temperature of the sun	Lecture and Assignment	CLO6

Halliday, D., Resnick, R., & Walker, J. (2001). *Fundamentals of physics* (6th ed.). John Willey and Sons, Inc.

Hossain, T. (1975). Text book of heat. Variety Books.

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Saha, M.N., & Srivastava, B.N. A. (1958) Treatise on heat. The Indian Press Private Ltd.

Sears, F.W. (1950). An Introduction to thermodynamics. Addison-Wesley Press.

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